

**CHIRAL TOLUENE-2, $\alpha$ -SULTAM AUXILIARIES:  
 ASYMMETRIC ALKYLATIONS, ACYLATIONS AND ALDOLIZATIONS OF N-ACYL DERIVATIVES**

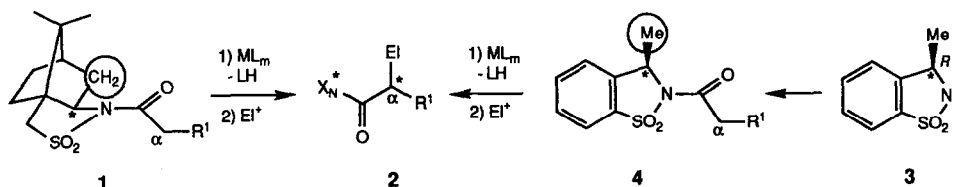
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**Abstract:** Successive treatment of (*R*)-*N*-acylsultams **4** with NHDMS/alkyl halides or NHDMS/acyl chlorides provides alkylated or acylated products **6** or **7**. Diastereoselective reductions of **7** with Zn(BH<sub>4</sub>)<sub>2</sub> or NaHB(*s*Bu)<sub>3</sub> gives "*syn*"- or "*anti*"-aldols **8** or **9**. Reaction of **4** with BEt<sub>3</sub>/TfOH/EtN(*i*Pr)<sub>2</sub> followed by addition of aromatic or aliphatic aldehydes affords diastereomerically pure "*syn*"-aldols **10**. Non-destructive removal of auxiliary **3** from **6**, **9** and **10** yields enantiomerically pure products **12** to **16**.

*N*-Acylbornane-10,2-sultams **1** provide stereochemically pure, crystalline  $\alpha$ -substituted products **2** via metalation and subsequent reaction with a variety of electrophiles E1<sup>+</sup> (e.g., alkyl halides, **1a,b** aldehydes, **1c** 1-chloro-1-nitrosocyclohexane, **1d** NBS, **1e** etc., Scheme 1).

Scheme 1



Exploring the potential of the related saccharin-derived sultam **3** (and its readily available antipode) as a chiral auxiliary,<sup>2</sup> we focus here on asymmetric carbon/carbon bond forming reactions. Sultam **3** was smoothly transformed into acyl derivatives **4** following protocols for the preparation of **1**.<sup>1</sup>

**Alkylations** (Scheme 2, Table 1):<sup>3</sup>

Deprotonation of **4** with sodium hexamethyldisilazide (NHMDS) followed by treatment with benzyl- or allyl iodide, flash chromatography (FC) and crystallization (MeOH) furnished stereochemically pure alkylation products **6a**, **6c** or **6f** (entries 1,3,6). Alkylations with less reactive methyl iodide or *t*-butyl  $\alpha$ -bromoacetate were carried out in the presence of HMPA or Bu<sub>4</sub>N<sup>+</sup>I<sup>-</sup> (entries 2,4 or 5) whereas ethyl iodide failed to give the corresponding product **6**.

Scheme 2

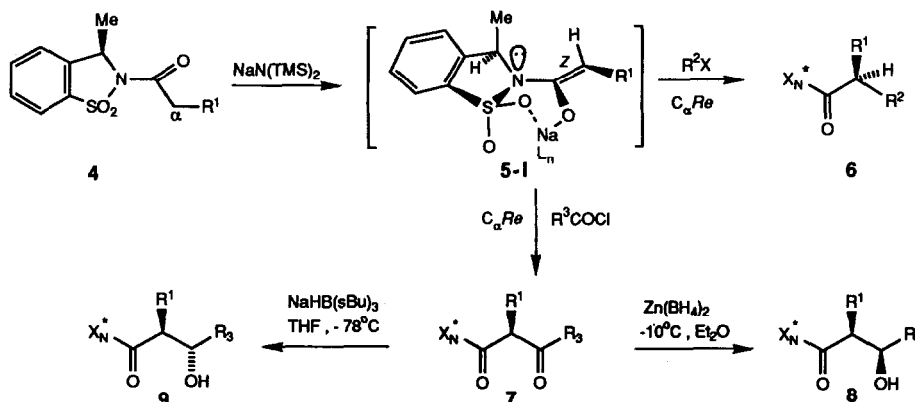


Table 1 : Asymmetric Alkylations **4** → **6**

	R <sup>1</sup>	R <sup>2</sup> X	Additive	Time [h]	Product	Purification	m.p.[°C]	d.e.[%]	yield [%]
1	Me	PhCH <sub>2</sub> I	--	2.5	<b>6a</b>	FC/cryst.	73-74	> 99	81
2	PhCH <sub>2</sub>	MeI	HMPA	6.5	<b>6b</b>	FC/cryst.	79-80	> 99	71
3	Me	CH <sub>2</sub> =CH-CH <sub>2</sub> I	--	3.5	<b>6c</b>	FC/cryst.	38-39	> 97 <sup>a</sup> )	75
4	CH <sub>2</sub> =CH-CH <sub>2</sub>	MeI	HMPA	5.5	<b>6d</b>	FC	oil	90 <sup>a</sup> )	91
5	Me	tBuO <sub>2</sub> C-CH <sub>2</sub> Br	Bu <sub>4</sub> NI	6.0	<b>6e</b>	FC	oil	> 99	76
6	N=CPh <sub>2</sub>	PhCH <sub>2</sub> I	--	8.0	<b>6f</b>	FC/cryst.	156-157	> 99	71

a) <sup>1</sup>H-NMR (360 MHz).

The topology of the transformations **4** → **6** are consistent with a predominant attack of the electrophile to the C(α)-Re face of the chelated (*Z*)-"enolate" **5-I**.

Acylations/Reductions (Scheme 2, Table 2): <sup>3</sup>

Table 2 : Asymmetric Acylation **4** → **7**

	R <sup>1</sup>	R <sup>3</sup>	d.e.[%] crude	Product	Purification	m.p.[°C]	d.e.[%]	yield [%]
7	Me	Ph	99.4	<b>7a</b>	FC/cryst.	120-121	> 99	89
8	Me	Me	86 <sup>a</sup> )	<b>7b</b>	FC/cryst.	104-105	99.2	62
9	Me	Et	91	<b>7c</b>	FC/cryst.	113-114	> 99	71
10	Me	CHMe <sub>2</sub>	92.8	<b>7d</b>	FC/cryst.	64-65	> 97	76
11	Me	CH <sub>2</sub> CHMe <sub>2</sub>	> 97 <sup>b</sup> )	<b>7e</b>	FC	oil	> 97 <sup>b</sup> )	81
12	Et	Ph	> 99	<b>7f</b>	FC/cryst.	108-109	> 99	71
13	Et	Me	97.6	<b>7g</b>	FC	oil	97.6	70

a) GC; b) <sup>1</sup>H-NMR (360 MHz): only 1 isomer visible.

The same topology was displayed by the C-acylation reactions **4** → **7**.<sup>4</sup> Successive treatment of acylsultams **4** with NHDMS and a carboxylic acid chloride afforded 1,3-dicarbonyl products **7**. Those epimerized only slowly with 1 M NEt<sub>3</sub> (25 molequiv in CH<sub>2</sub>Cl<sub>2</sub>, r.t., 12 h: (*S*)-**7a** → (*S*)-**7a** + (*R*)-**7a** 1:1), analogous to *N*-(3-oxoacyl)oxazolidinones.<sup>4a</sup> The lowest diastereoselectivity was observed on acylation of **4**, R<sup>1</sup> = Me with (the sterically least demanding) acetyl chloride (entry 8). However, resulting **7b** (d.e. = 86%) was conveniently purified (d.e. = 99.2%) by FC/crystallization, as were most other C-acylation products **7**.

Table 3 : *Syn*- or *Anti*- Aldols **8** or **9** by Diastereoselective Reductions of *N*-(3-Oxoacyl)sultams **7**

	R <sup>1</sup>	R <sup>3</sup>	Hydride	Ratio 8/9	Purification	Major Product	m.p.[°C]	purity[%]	yield [%]
14	Me	Ph	Zn(BH <sub>4</sub> ) <sub>2</sub>	99.1 : 0.9	cryst.	<b>8a</b>	126-127	> 99	82
15	Me	Me	Zn(BH <sub>4</sub> ) <sub>2</sub>	91.3 : 8.7	FC	<b>8b</b>	oil	> 99	81
16	Me	CHMe <sub>2</sub>	Zn(BH <sub>4</sub> ) <sub>2</sub>	98.8 : 1.2	cryst.	<b>8d</b>	108-109	> 99	72
17	Me	Ph	NaHB(sBu) <sub>3</sub>	0.7 : 99.3	FC	<b>9a</b>	146-147	> 99	82
18	Me	Me	NaHB(sBu) <sub>3</sub>	0.2 : 99.8	FC	<b>9b</b>	oil	> 99	80
19	Me	CH <sub>2</sub> CHMe <sub>2</sub>	NaHB(sBu) <sub>3</sub> <sup>b</sup> )	2.1 : 97.9	FC	<b>9c</b>	oil	> 99	53
20	Et	Me	NaHB(sBu) <sub>3</sub>	--	FC	<b>9g</b>	oil	> 97 <sup>a</sup> )	80

a) <sup>1</sup>H-NMR (200 MHz): only 1 isomer visible; b) at -50°C.

Chelate-controlled reduction of *N*-(3-oxoacyl)sultams **7** with zinc borohydride **4a**, **5a** proceeded without epimerization and afforded selectively "*syn*"-aldols **8** (Scheme 2, Table 3, entries 14 to 16).

"Anti"-aldols **2** were obtained almost exclusively when reducing the keto group of **7** with sodium tri-*s*-butylborohydride (entries 17 to 20).<sup>5</sup> The purity of aldols **8** and **9** was increased to >99% by crystallization or FC.

### Aldolizations (Scheme 3, Table 4):<sup>3</sup>

More directly, "syn"-aldols **10** were produced by successive treatment of acylsultam **4** ( $R^1 = \text{Me}$ ) with a dialkylboryl triflate/EtN(*i*Pr)<sub>2</sub> and an aldehyde (Scheme 3, Table 4).

#### Scheme 3

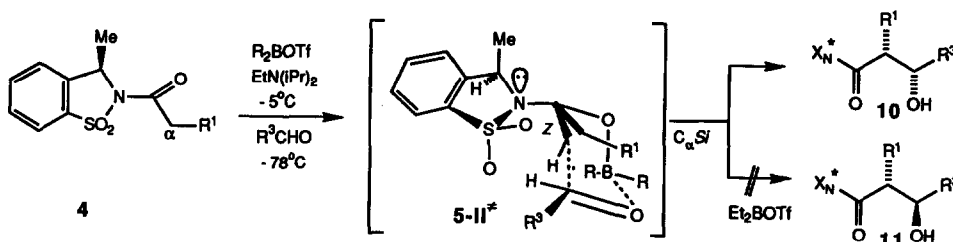


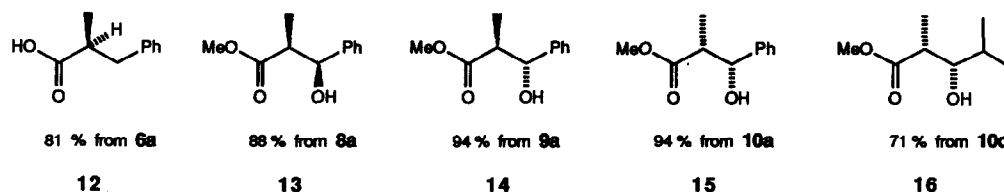
Table 4 : Asymmetric Aldolizations **4**  $\rightarrow$  **10**

	R <sup>1</sup>	Aldehyde R <sup>2</sup>	Boryltriflate R (molequiv)	Enolate formation [°C]	Conversion of <b>4</b>	Product Ratio 10/8/11/9	Product	Yield cryst [%]	M.p.[°C]	Purity [%]
21	Me	Ph	Bu (1.4)	-10 to -5	94	97.8/ 0 / 2.2 / 0	<b>10a</b>	58	117-118	> 99
22	Me	Ph	Bu (2.0)	-10 to -5	62	7.5/41.4/27.7/23.4	--	--	--	--
23	Me	<i>i</i> Bu	Bu (0.7)	-5	56	98 / 2 / 0 / 0	--	--	--	--
24	Me	<i>i</i> Bu	Bu (1.2)	-5	61	21 / 43 / 36 / 0	--	--	--	--
25	Me	<i>i</i> Pr	Bu (1.6)	-5	90	93.7/ 1.5/ 1.8/ 3	<b>10d</b>	50	112-114	98.5
26	Me	Ph	Et (2.0) <sup>a</sup>	-5	100	>99 / 0 / 0 / 0	<b>10a</b>	84	118-119	> 99
27	Me	Me	Et (2.0) <sup>a</sup>	-5	100	>99 / 0 / 0 / 0	<b>10b</b>	71	104-106	> 99
28	Me	<i>i</i> Pr	Et (2.0) <sup>a</sup>	-5	99	>99 / 0 / 0 / 0	<b>10d</b>	95 <sup>b</sup>	oil	> 99
29	Me	<i>i</i> Bu	Et (2.0) <sup>a</sup>	-5	92	>99 / 0 / 0 / 0	<b>10e</b>	78	113-114	> 99

a) Prepared *in situ* from  $\text{BEt}_3$  and  $\text{CF}_3\text{SO}_3\text{H}$ ; b) Purified by FC.

Employing freshly prepared dibutylboryl triflate led to incomplete conversions and an excess of  $\text{Bu}_2\text{BOTf}/\text{EtN}(i\text{Pr})_2$  resulted in lower stereoselectivities (cf., entries 21/22, 23/24). More conveniently, more efficiently and more selectively, aldols **10** were obtained by using *in situ* prepared diethylboryl triflate/ $\text{EtN}(i\text{Pr})_2$  (2 molequiv, entries 26 to 29).<sup>1c</sup> Under these conditions a variety of aldehydes furnished without exception pure aldols **10** in high yields. As expected,<sup>1c</sup> aldolizations **4**  $\rightarrow$  **10** reflect an electrophilic attack to the opposite "enolate"  $\pi$ -face (C $\alpha$ -Si) than observed with alkylations **4**  $\rightarrow$  **6** and acylations **4**  $\rightarrow$  **7**. We ascribe this dichotomy to the transition state **5-II\*** again involving a (Z)-"enolate". However, with the boron atom being fully coordinated (and, thus, incapable of chelation with a  $\text{SO}_2$  oxygen atom) the "enolate" adopts an (electrostatically favored) N- $\text{SO}_2/\text{C-OML}_n$  *s-trans* conformation.

### Hydrolyses. Absolute Configurations (Scheme 4):<sup>3</sup>



Hydroperoxide-assisted saponification (**6a** → **12**), combined with esterification (**8a** → **13**, **9a** → **14**, **10a** → **15** and **10d** → **16**) provided enantiomerically pure  $\alpha$ -branched carboxylic acid **12** or methoxycarbonyl aldols **13** to **16** in good yields with 92 to 94% recovery of auxiliary sultam **3**. The absolute configurations of **12**, **13**, **15** and **16** were determined by comparing their optical rotations with published reference values. This correlation allowed to assign the sense of induction in alkylations **4** → **6**, acylations **4** → **7** and **4** → **10** as depicted in the Schemes 2, 3 and 4.

#### Conclusion:

The topicities of the alkylations, acylations and aldolizations described here parallel those observed with *N*-acylbornane-10,2-sultams **1**.<sup>1</sup> An advantageous application of a related toluene-2, $\alpha$ -sultam auxiliary will be reported in due course.

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#### REFERENCES AND NOTES

- 1) a) W. Oppolzer, R. Moretti, S. Thomi, *Tetrahedron Lett.* **1989**, *30*, 5603; b) *idem, ibid.*, **1989**, *30*, 6009; c) W. Oppolzer, J. Blagg, I. Rodriguez, E. Walther, *J. Am. Chem. Soc.* **1990**, *112*, 2767; d) W. Oppolzer, O Tamura, *Tetrahedron Lett.* **1990**, *31*, 991; e) W. Oppolzer, *Pure & Appl. Chem.* **1990**, *62*, in press.
- 2) W. Oppolzer, M. Wills, C. Starkemann, G. Bernardinelli, *Tetrahedron Lett.* **1990**, *31*, in press.
- 3) All new compounds were characterized by IR, <sup>1</sup>H-NMR, <sup>13</sup>C-NMR and MS. D.e. values were determined by direct HPLC analyses, unless otherwise stated. Aldol-stereoisomer ratios **10/8/11/2** were determined by comparison (HPLC of crude products or GC of TBDMS derivatives) with authentic mixtures. The <sup>1</sup>H-NMR spectra showed typical coupling constants  $J(2,3) = 3-4.5$  Hz for "syn"-aldols (**8**, **10**, **13**, **15**, **16**) and  $J(2,3) = 7-8.5$  Hz for "anti"-aldols (**9**, **14**). Melting point: **4**, R<sup>1</sup> = Me: 94.5-95°C.  $[\alpha]_D$  values, (20° C, CHCl<sub>3</sub>, g/100 ml): **4**, R<sup>1</sup> = Me: -19.5° (1.2); **12**: +28.3°(0.85), Lit <sup>1a</sup>: +29.3° (1.15); **13**: -23.6° (1.66), Lit <sup>1c</sup> antipode: +23.5° (3.2); **14**: +58.9° (1.88); **15**: +26.1°, Lit <sup>1c</sup> : +23.5° (3.2); **16**: +9.4° (0.83), Lit <sup>1c</sup> : -7.14° (1.19).  
The following procedures are representative: *Acylation of Sultam 3*: Propionyl chloride (2.28 ml, 26 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 ml) was added to a mixture of sultam **3** (4.0 g, 22 mmol), NEt<sub>3</sub> (34 ml, 24 mmol) and CH<sub>2</sub>Cl<sub>2</sub> (22 ml) at 0°C. Stirring for 3 h, workup and crystallization (hexane/Et<sub>2</sub>O) gave **4**, R<sup>1</sup> = Me (4.48 g, 85%). Acylsultam **4**, R<sup>1</sup> = N=CPh<sub>2</sub> was prepared from **3**, MeO<sub>2</sub>CCH<sub>2</sub>N=CPh<sub>2</sub>/Me<sub>3</sub>Al c.f. ref <sup>1b</sup>. *Alkylation*: Sultam **4**, R<sup>1</sup> = Me (250 mg, 1.04 mmol) in THF (3 ml) was added to 1 M NHDMS (THF, 1.14 mmol) over 0.5 h at -78°C. Stirring for 0.5 h, addition of benzyl iodide (270  $\mu$ l, 2.14 mmol), stirring at -78°C for 2.5 h, quenching with sat. aq. NH<sub>4</sub>Cl, workup, FC (hexane/EtOAc 9:1) and crystallization (MeOH) gave product **6a** (277 mg, 81%). *Acylation*: 1 M NHDMS (2.46 ml in THF) was added to sultam **4**, R<sup>1</sup> = Me (535 mg, 2.24 mmol) in THF (10 ml) at -78°. Stirring for 0.5 h, addition of benzoyl chloride (0.68 ml, 3.3 mmol) at -78°C, continued stirring for 0.5 to 1h, quenching with aq. phosphate buffer (pH = 7), work up and crystallization (EtOH) provided *N*-(3oxoacyl)sultam **7a** (668 mg, 89%). (Acylation of **4** R<sup>1</sup> = Me with acetyl- and *n*-propionyl chlorides, entries **8,9** were carried out in the presence of HMPA (1 molequiv)). *Reductions of Oxoacylsultam*: a) 0.16 M Zn(BH<sub>4</sub>)<sub>2</sub> (Et<sub>2</sub>O, 2 ml) was added to **7a** (100 mg, 0.29 mmol) in Et<sub>2</sub>O (10 ml) at -10°. Stirring at -10°C for 0.5 h, addition of 3N aq. HCl, workup and crystallization (hexane/ether) gave "syn"-aldol **8a** (82 mg, 82%). b) 1 M Sodium selectride (THF, 1 ml) was added dropwise to **7a** (250 mg, 0.73 mmol) in THF (4 ml) at -78°C. Stirring for 0.5 h, quenching with phosphate buffer, work up and FC (SiO<sub>2</sub>, hexane/EtOAc 4:1) yielded "anti"-aldol **9a** (205 mg, 82%). *Aldolization*: CF<sub>3</sub>SO<sub>3</sub>H (73.8  $\mu$ l, 0.84 mmol) was added to 1M BEt<sub>3</sub> (hexane, 0.84 mmol) at r.t.. Stirring at 40°C until the gas evolution had ceased (~5 min), cooling to -5°C, addition of **4**, R<sup>1</sup> = Me (100 mg, 0.42 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (1.4 ml) and of 1 M NEt(iPr)<sub>2</sub> (CH<sub>2</sub>Cl<sub>2</sub>, 0.9 ml), stirring at -5°C for 0.5 h, addition of benzaldehyde (85 $\mu$ l, 0.84 mmol) at -78°C, stirring at -78°C for 20 min, addition of aq. phosphate buffer (2 ml), extraction with Et<sub>2</sub>O, washing of the extracts with sat. aq. NH<sub>4</sub>Cl, drying (MgSO<sub>4</sub>), evaporation, FC (SiO<sub>2</sub>, hexane/EtOAc 2:1) and crystallization (MeOH) afforded aldol **10a** (122mg, 84%). *Saponification/Esterification*: Aq. (30%) H<sub>2</sub>O<sub>2</sub> (74  $\mu$ l, 0.72 mmol) and LiOH.H<sub>2</sub>O (15.2 mg, 0.36 mmol) were added at 0°C to **10a** (100 mg, 0.29 mmol) in THF/H<sub>2</sub>O (3:1, 2 ml). Stirring at 0°C for 2 h, addition of sat. aq. Na<sub>2</sub>SO<sub>3</sub> (1 ml), basification to pH = 11 (NH<sub>4</sub>OH), extraction (CH<sub>2</sub>Cl<sub>2</sub>), drying and evaporation of the extracts and crystallization of the residue gave sultam **3** (49 mg, 92%). Acidification of the aq. phase to pH = 1 (HCl), extraction (Et<sub>2</sub>O), drying of the extracts, addition of CH<sub>2</sub>N<sub>2</sub> (excess), evaporation and FC (hexane/EtOAc 3:1) furnished hydroxyester **15** (53 mg, 94%).
- 4) Previously published asymmetric acylations of chiral "enolates": a) D.A. Evans, M.D. Ennis, T. Le, N. Mandel, G. Mandel *J. Am. Chem. Soc.* **1984**, *106*, 1154; b) Y. Ito, T. Katsuki, M. Yamaguchi, *Tetrahedron Lett.* **1984**, *25*, 6015.
- 5) Previously published reductions of 3-oxoamide derivatives a) with Zn(BH<sub>4</sub>)<sub>2</sub> giving "syn"-aldols **4**; b) with KBET<sub>3</sub>H giving "anti"-aldols: Y. Ito T. Katsuki, M. Yamaguchi, *Tetrahedron Lett.* **1985**, *26*, 4643.

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